Pressure Stimulated Charge-Crossover in Transition-Metal Oxides and Hydroxides

<u>Moshe P. Pasternak</u>, Ella Milner, Weiming Xu, and Gregory Kh. Rozenberg School of Physics & Astronomy, Tel Aviv University, 69978 Tel Aviv, ISRAEL

Mössbauer spectroscopy, x-ray diffraction (XRD), and electrical resistance (R(P,T)) studies were applied to investigate the high properties of the Mott insulator antiferromagnetic layered Fe(OH)₂ and CuFeO₂. In Fe(OH)₂ an unforeseen process by which a gradual Fe²⁺ oxidation takes place, starting at ~8 GPa reaching 70% Fe³⁺-abundance at 40 GPa. Based on XRD and R(P,T) data it is unequivocally concluded that this non-reversible process, Fe²⁺ \rightarrow Fe³⁺ + e⁻, results in Fe²⁺ converting into Fe³⁺ with no structural transition. The "ousted" electrons form a deep band within the Fe(OH)₂ high-pressure electronic-manifold becoming weakly-localized at P > 50 GPa. This process is attributed to an effective ionization potential created by the pressure-induced orientation deformed (OH)⁻¹ dipoles and the unusual small binding energy of the valence electron in Fe²⁺(OH)₂.

Magnetic properties of the 2D spin-frustrated CuFeO₂ delafossite have been studied up to 100 GPa. The partially disordered spin arrangement at ambient pressure in the 11 – 16 K range, transforms with pressure to a long-range ordered "5-sublattice" phase with a distinct T_N, a similar role played by external magnetic field in neutron studies. This phase gradually substitutes for the "4-sublattice" magnetic ground state present at ambient pressure, reaching 100% at 19 GPa. The twofold increase of T_N at 19 GPa is explained in terms of the unusual increase of the intra-planar direct exchange $J_{||}$ caused by the anomalous anisotropic compression of CuFeO₂ in which *c/a* increases with pressure. With further pressure increase at ~ 30 GPa, about half of the Fe³⁺ and Cu¹⁺ ions undergo a *reduction-oxidation*1 transition: (Fe³⁺,Cu¹⁺) \xrightarrow{P} (Fe²⁺,Cu²⁺) explained by pressure-induced *band overlap*. New magnetic sublattices are formed composed of Fe²⁺(S=2) and Cu²⁺(S=1/2) enhancing dramatically the T_N. At 50 GPa both Fe ions undergo a spin transition and finally at 80 GPa a *Mott* transition into a metallic state takes place.